

**[Indenyllithium·2(*N,N,N'*-trimethylethylene-  
diamine)]<sub>∞</sub>: N–H hydrogen bridges to the indenyl  
anion****Gernot Boche, Burkhard Ledig,  
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GermanyCorrespondence e-mail:  
harms@chemie.uni-marburg.de**Key indicators**Single-crystal X-ray study  
*T* = 173 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$   
Disorder in solvent or counterion  
*R* factor = 0.040  
*wR* factor = 0.114  
Data-to-parameter ratio = 9.7For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.

The crystal structure of the title compound, bis(*N,N,N'*-trimethylethylenediamine)lithium indenide,  $[\text{Li}(\text{C}_5\text{H}_{14}\text{N}_2)_2]-(\text{C}_9\text{H}_7)$ , forms discrete layers of cations and anions, with N–H···C hydrogen-bond interactions between these layers. Both the cation and anion exhibit a crystallographic centre of inversion. This leads to a twofold disorder of the anion.

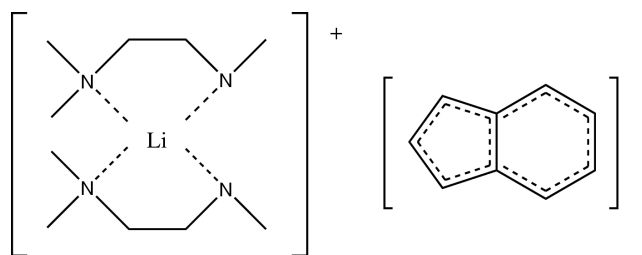
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**Comment**

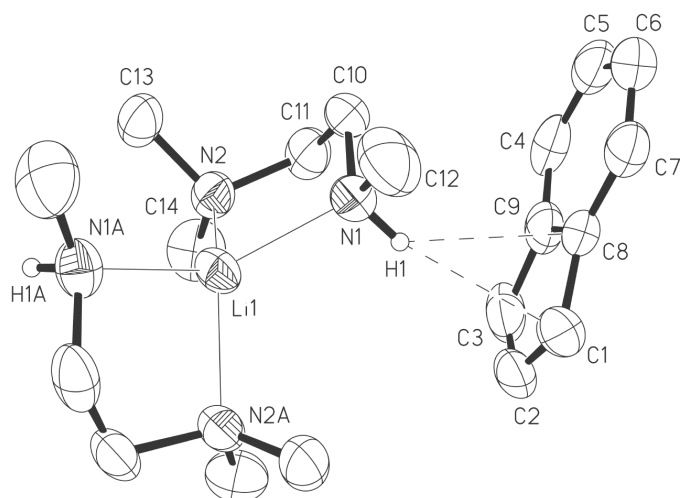
Although hydrogen-bond bridges to ‘carbanions’ have been extensively studied in solution (Cram, 1965; Ford, 1970; Pascault & Gole, 1971; Hogen-Esch, 1973; Hussénium *et al.*, 1989; Eliasson *et al.*, 1990; McEwen & Ahlberg, 1992; McEwen *et al.*, 1993), there are only a few cases of such bonds observed in the solid state (Laube *et al.*, 1985; Klebe *et al.*, 1987; Buchholz, Harms, Marsch *et al.*, 1989; Buchholz, Harms, Massa & Boche, 1989; Armstrong *et al.*, 1991; Lambert *et al.*, 1992). It was also recognized that the interaction of  $\text{Li}^+$  with, *e.g.*, the N atom of an amine  $\text{RNH}_2$  leads to better hydrogen-bonding properties for the amine (Buchholz, Harms, Marsch *et al.*, 1989). Here we report the crystal structure of [indenyllithium·2(*N,N,N'*-trimethylethylenediamine)]<sub>∞</sub>. The indenyl ‘anion’ has been of special interest for hydrogen bonding with a ‘carbanion’, and for mechanistic studies of H/D exchange reactions (Cram, 1965; Eliasson *et al.*, 1990; Hussénium *et al.*, 1989; McEwen & Ahlberg, 1992).



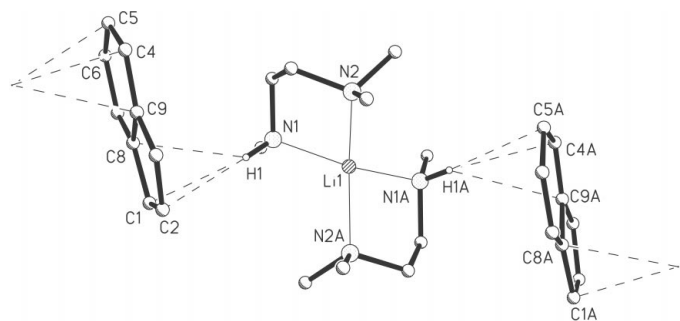
(I)

**Experimental**

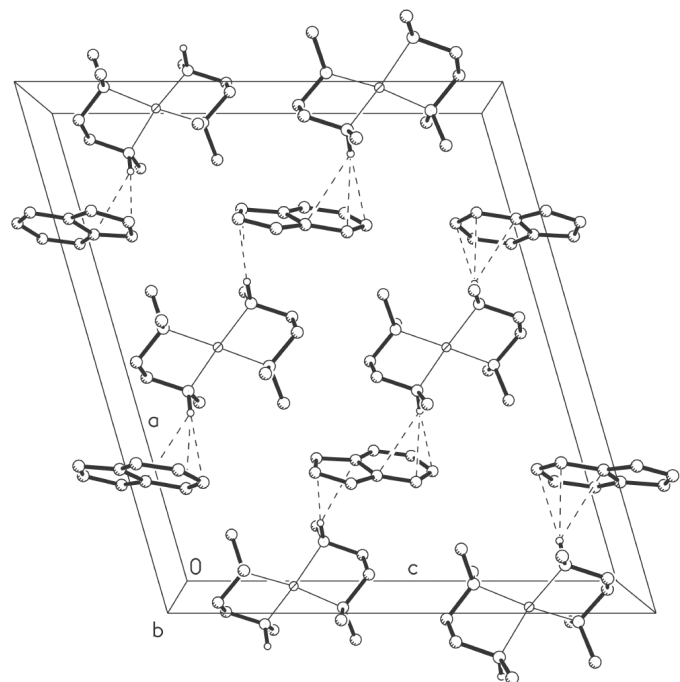
0.49 mmol of LDA (lithium diisopropylamide) in 5 ml THF was treated with 0.46 mmol indene at 195 K. The yellow solution was warmed to room temperature and 3.14 mmol TriMEDA (trimethylethylenediamine) was added. After several days at room temperature, yellow–brown single crystals were obtained.



**Figure 1**  
The structure of the title compound. Displacement ellipsoids are drawn at the 50% probability level. The disorder is not shown.



**Figure 2**  
N—H...C interactions in the title compound.



**Figure 3**  
Packing diagram, viewed down the *b* axis.

### Crystal data

[Li(C<sub>5</sub>H<sub>14</sub>N<sub>2</sub>)<sub>2</sub>](C<sub>9</sub>H<sub>7</sub>)  
*M<sub>r</sub>* = 326.45  
 Monoclinic, *C2/c*  
*a* = 17.669 (8) Å  
*b* = 7.921 (2) Å  
*c* = 15.598 (4) Å  
 $\beta$  = 106.10 (1)°  
*V* = 2097.4 (12) Å<sup>3</sup>  
*Z* = 4

*D<sub>x</sub>* = 1.034 Mg m<sup>-3</sup>  
 Cu *K*α radiation  
 Cell parameters from 25 reflections  
 $\theta$  = 39.8–46.9°  
 $\mu$  = 0.46 mm<sup>-1</sup>  
*T* = 173 (2) K  
 Prism, colourless  
 0.40 × 0.25 × 0.20 mm

### Data collection

Enraf–Nonius CAD-4 diffractometer  
 $\omega$  scans  
 2624 measured reflections  
 1323 independent reflections  
 1111 reflections with *I* > 2σ(*I*)  
*R<sub>int</sub>* = 0.058

$\theta_{\max}$  = 55.0°  
*h* = -18 → 18  
*k* = -8 → 8  
*l* = 0 → 16  
 3 standard reflections  
 frequency: 60 min  
 intensity decay: none

### Refinement

Refinement on *F*<sup>2</sup>  
*R*[*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.040  
*wR*(*F*<sup>2</sup>) = 0.114  
*S* = 1.07  
 1323 reflections  
 137 parameters  
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0452P)^2 + 0.8262P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.11 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.14 \text{ e \AA}^{-3}$   
 Extinction correction: *SHELXL97*  
 Extinction coefficient: 0.0016 (2)

**Table 1**

Hydrogen-bonding geometry (Å, °).

<i>D</i> — <i>H</i> ... <i>A</i>	<i>D</i> — <i>H</i>	<i>H</i> ... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> — <i>H</i> ... <i>A</i>
N1—H1...C1	0.892 (19)	2.522 (19)	3.386 (7)	163.0 (15)
N1—H1...C4 <sup>i</sup>	0.892 (19)	2.553 (19)	3.373 (6)	153.0 (14)
N1—H1...C5 <sup>i</sup>	0.892 (19)	2.830 (18)	3.721 (9)	176.5 (14)
N1—H1...C8	0.892 (19)	2.648 (18)	3.359 (5)	137.3 (13)
N1—H1...C9 <sup>i</sup>	0.892 (19)	2.738 (18)	3.400 (5)	131.9 (13)

Symmetry code: (i)  $\frac{3}{2} - x, \frac{1}{2} - y, -z$ .

The structure was initially solved in space group *Cc*. Displacement-ellipsoid plots after the anisotropic refinement indicated clearly a disorder of the indenyl anion due to inversion symmetry. The refinement was continued in space group *C2/c*, in which the anion adopts crystallographic inversion symmetry leading to a twofold disorder. A complete indenyl anion (occupancy factor 0.5) has been modelled using restraints for bond distances, planarity and displacement parameters. The derived geometry (C—C bond distances 1.37–1.43 Å) has been fixed as a rigid group in the final stage of the refinement. A similar disorder of an indenyl anion has been found in the crystal structure of bis[bis(dimethylphosphino)ethane-*P,P'*]-rhodium indenide (Marder & Williams, 1987). The N-bonded H atom was located and allowed to refine isotropically.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD4* (Harms, 1997); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Sheldrick, 1996); software used to prepare material for publication: *SHELXTL*.

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